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Akihiko Namba

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EXAMINER

HUBER, ROBERT T

ART UNIT

PAPER NUMBER

2892

MAIL DATE

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09/03/2009

PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary	Application No. 10/580,346	Applicant(s) NAMBA ET AL.	
	Examiner ROBERT HUBER	Art Unit 2892	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 02 June 2009.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1-16 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1-16 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☒ The drawing(s) filed on 25 May 2006 is/are: a) ☒ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☒ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☒ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☒ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|------------------------------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------------|
| 1) <input type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413)
Paper No(s)/Mail Date. _____ |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08)
Paper No(s)/Mail Date _____ | 6) <input type="checkbox"/> Other: _____ |

DETAILED ACTION

Claim Rejections - 35 USC § 112

1. The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

2. Claims 1 – 11 and 13 – 16 are rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the enablement requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to enable one skilled in the art to which it pertains, or with which it is most nearly connected, to make and/or use the invention. In particular, claims 1 and 15 recite *“an electron concentration of said first diamond semiconductor exhibits a negative correlation with temperature, in a temperature range having a width of 100°C or more and included within a temperature region from 0°C to 300°C”*, which is a property of the device, however the Applicants have not provided a substantial disclosure to support such a claim. Although the Applicants have recited in the specification numerous times, as well as in the remarks filed on December 9, 2008, that the device has a single layer with the claimed properties, the specification does not indicate how or why such properties exist. In particular, it is well-known to one of ordinary skill in the art that the electron concentration of a semiconductor exhibits a positive correlation with temperature. In other words, as temperature increase, the density of electrons (carriers) increases monotonically with temperature. The Applicants claim the properties of a "new" material without adequately describing the new material and accounting for such properties.

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Hence, the claimed invention does not enable one of ordinary skill in the art to make or use such an invention, since the claimed invention appears to violate known physical laws. Claims 2 – 11 and 13 depend on claim 1, and claim 16 depends on claim 15.

It appears from the specification that the existence of a second semiconductor layer allows the transfer of electrons from the first semiconductor layer to the second semiconductor layer, thereby decreasing the concentration of electrons in the first semiconductor layer, which does not appear to violate known physical laws (as supported in ¶ [0050] of the Specification). Hence, claim 12, which recites the second semiconductor layer, is not rejected under 35 USC 112, first paragraph.

Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

4. Claims 1 – 6, 8, 11, and 13 are rejected under 35 U.S.C. 102(b) as anticipated by or, in the alternative, under 35 U.S.C. 103(a) as obvious over Imai et al. (US 5,001,452, prior art of record).

a. Regarding claim 1, **Imai discloses a diamond n-type semiconductor** (e.g. Example 1, being in col. 4, line 66) **comprising a first diamond semiconductor which has n-type conduction** (col. 4, line 68 – col. 5, line discloses forming an n-type diamond semiconductor layer, doped with sulfur, on

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a diamond substrate) **and in which a distortion is artificially formed** (as disclosed in Imai, the n-type layer is doped with dopants (S), and therefore a distortion is formed in the diamond semiconductor lattice due to the sulfur impurity. Furthermore, the patentability of a product does not depend on the method of production. See MPEP 2113).

wherein said first diamond semiconductor has an n-type dopant concentration adjusted by vapor-phase growth (col. 5, line 3 discloses forming the n-type dopant concentration layer by CVD, which is a vapor phase deposition. Furthermore, the patentability of a product does not depend on the method of production. See MPEP 2113.) **such that an electron concentration of said first diamond semiconductor exhibits a negative correlation with temperature, in a temperature range having a width of 100°C or more and included within a temperature region from 0°C to 300°C (see below).**

b. Claim 2, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, wherein said first diamond semiconductor has a Hall coefficient exhibiting a positive correlation with temperature, in the temperature range**

c. Claim 3, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, wherein the temperature range included within**

the temperature region from 0°C to 300°C has a width of over 200°C or more (see below)

d. Claim 4, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, wherein said first diamond semiconductor has a resistivity of 500 Ωcm or less at a temperature within the temperature region from 0°C to 300°C (see below)**

e. Claim 5, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, wherein the electron concentration of said first diamond semiconductor is always 10^{16} cm^{-3} or more in the temperature region from 0°C to 300°C (see below).**

Regarding claims 1 – 5, the device of Example 1 of Imai contains an n-type diamond semiconductor layer containing a Sulfur dopant concentration, as disclosed in col. 2, lines 30 – 32 and Table 1, which resides on a diamond substrate, as disclosed in col. 5, lines 1 - 2. Since the device of Imai meets the structural limitations of the claimed invention of the Applicant, the properties of the applicant's invention, such as the temperature dependence of the electron concentration and Hall coefficient as claimed in claims 1 - 5, are presumed inherent to the device of Imai. See MPEP 2112.01.

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- f. Regarding claim 6, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, wherein said first diamond semiconductor contains more than $5 \times 10^{19} \text{ cm}^{-3}$ in total of at least one kind of donor element** (col. 2, lines 30 – 32 and Table 1 disclose a dopant concentration to be between $1 \times 10^{12} \text{ cm}^{-3}$ and $1 \times 10^{21} \text{ cm}^{-3}$, and therefore anticipates the claimed value).
- g. Regarding claim 8, **Imai discloses a diamond n-type semiconductor according to claim 6, as cited above, wherein said first diamond semiconductor contains at least S (sulfur) as the donor element** (e.g. as disclosed in col. 4, line 68).
- h. Regarding claim 11, **Imai discloses a diamond n-type semiconductor according to claim 1, wherein said first diamond semiconductor is monocrystal diamond** (e.g. col. 2, line 28 discloses the formation of the first diamond semiconductor by "single-crystal" growth).
- i. Regarding claim 13, **Imai discloses a semiconductor device at least partly employing a diamond n-type semiconductor according to claim 1** (as disclosed in col. 4, lines 38 - 44)

Claim Rejections - 35 USC § 103

5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

6. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

7. Claim 7 is rejected under 35 U.S.C. 103(a) as being unpatentable over Imai in view of Hasegawa et al. (US 2002/0127405 A1, prior art of record). **Imai discloses the diamond semiconductor according to claim 6, as cited above, but is silent with respect to explicitly disclosing said first diamond semiconductor contains at least P (phosphorus) as the donor element. However, Imai does acknowledge that phosphorus may be used as a dopant in semiconductors (col. 2, lines 20 - 21).**

Hasegawa discloses that both phosphorus and sulfur can be used as n-type dopants in diamond semiconductors (e.g. ¶ [0032] discloses the use of sulfur, and ¶ [0037] discloses that phosphorus may also be used).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the device of Imai such that phosphorus was used as the dopant in the n-type diamond semiconductor layer since Imai discloses that sulfur is used as the n-type dopant, and Hasegawa discloses that both sulfur and phosphorus may be used as the n-type dopant in the diamond semiconductor. One would have been motivated to use phosphorus instead of sulfur in order to adjust the band-gap and electrical properties of the semiconductor to optimize the semiconductor properties, such as conductivity, for a desired circuit, as discussed by Hasegawa (¶ [0037]).

8. Claims 9, 14, and 15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Imai in view of Yoshida (US 6,340,393 B1, prior art of record).

a. Regarding claim 9, **Imai discloses a diamond n-type semiconductor according to claim 1, as cited above, however Imai is silent with respect to said first diamond semiconductor containing an impurity element other than the donor element together with the donor element.**

Yoshida discloses a combining a second impurity element together with the donor element in a diamond semiconductor (col. 2, lines 51 – 52, col. 5, lines 59 – 61, and col. 5, lines 65 - 67).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the device of Imai to include a second impurity element along with the donor element, since it was known that one may combine second impurity elements into doped diamond semiconductor materials, as taught by Yoshida, because it would allow a stabilization of the semiconductor layer with a large dopant density (as discussed in Yoshida, col. 2, lines 51 - 53).

b. Regarding claim 14, **Imai discloses the diamond n-type semiconductor according to claim 1, as cited above, but is silent with respect to the device being used in at least an electron emitting part of an electron emitting device.**

Yoshida discloses that diamond semiconductor devices can be used as an electron emitter (col. 5, lines 18 – 19).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to use the diamond semiconductor of Imai as an electron emitter since Yoshida discloses that such semiconductor devices can be used as electron emitters. One would be motivated to use the devices in such a manner since a low resistivity exists in such devices, creating an efficient electron emitter.

c. Regarding claim 15, **Imai discloses a method of manufacturing a diamond n-type semiconductor (e.g. Example 1, being in col. 4, line 66), said method comprising the steps of:**

preparing a diamond substrate (substrate disclosed in col. 5, line 1);
and
epitaxially growing a diamond semiconductor on said diamond substrate by vapor phase growth (col. 4, line 68 – col. 5, line 3 discloses forming the diamond semiconductor by CVD (chemical vapor deposition) and col. 2, lines 25 – 26 disclose forming the diamond semiconductor by vapor phase growth. Col. 5, line 8 discloses the diamond semiconductor to be epitaxial),
whereby said diamond semiconductor has n-type conduction (e.g. as disclosed in col. 5, lines 7 - 10) **and has a distortion which is artificially formed therein** (as disclosed in Imai, the n-type layer is doped with dopants (S), and therefore a distortion is formed in the diamond semiconductor lattice due to the impurity),

wherein said diamond semiconductor has an n-type dopant concentration adjusted by said vapor-phase growth (col. 4, line 68 – col. 5, line 3 discloses forming the diamond semiconductor by CVD (chemical vapor deposition) and col. 2, lines 25 – 26 disclose forming the diamond semiconductor by vapor phase growth. Table 1 discloses adjusting the n-type dopant concentration). **Imai is silent with respect to disclosing artificially introducing an impurity element other than a donor element to said diamond substrate while growing the diamond semiconductor.**

Yoshida discloses artificially introducing an impurity element other than a donor element to a diamond substrate while growing the diamond

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semiconductor (e.g. col. 2, lines 51 – 52, col. 5, lines 59 – 61, and col. 5, lines 65 - 67).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the method of Imai such that an impurity element other than the donor (dopant) element is introduced to the diamond substrate while growing the diamond semiconductor since Yoshida discloses that the addition of impurities while growing diamond semiconductors on diamond substrates can be advantageous. One would have been motivated to add the impurity to the doped diamond semiconductor because it would aid in the stabilization of the diamond semiconductor layer with a large dopant density (as discussed in Yoshida, col. 2, lines 51 - 53).

Regarding the claimed limitation of the electron concentration of said diamond semiconductor exhibiting a negative correlation with temperature, in a temperature range having a width of 100°C or more and which included within a temperature region from 0°C to 300°C, the method of Example 1 of Imai (comprising forming an n-type diamond semiconductor layer containing a concentration of a Sulfur dopant, which resides on a diamond substrate), in view of the method of Yoshida (introducing an impurity element to the to the diamond substrate while growing a doped diamond semiconductor layer), is an obvious teaching over the method of claim 15. Since the method of Imai in view of Yoshida results in the structural limitations of the claimed method invention of the Applicant, the properties of the applicant's invention, such as the

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temperature dependence of the electron concentration, are presumed inherent to the device of Imai in view of Yoshida. See MPEP 2112.01).

9. Claims 10 and 16 are rejected under 35 U.S.C. 103(a) as being unpatentable over Imai in view of Yoshida as applied to claim 9 above, and further in view of Hasegawa.

a. Regarding claim 10, **Imai in view of Yoshida disclose a diamond n-type semiconductor according to claim 9, as cited above, however they are silent with respect to explicitly disclosing said first diamond semiconductor containing at least $1 \times 10^{17} \text{ cm}^{-3}$ of Si as the impurity element.**

Hasegawa discloses that a concentration of 1×10^{16} to $1 \times 10^{21} \text{ cm}^{-3}$ of silicon can be used as an impurity element when doping semiconductor diamond (paragraphs [0037] – [0038]).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the device of Imai in view of Yoshida, such that the other impurity element is silicon with a concentration of 1×10^{16} to $1 \times 10^{21} \text{ cm}^{-3}$ since Hasegawa discloses that silicon can be used in such concentrations to dope semiconductor diamond with a p-type material, and Yoshida discloses that a combination of n-type and p-type materials may be doped into a diamond semiconductor in order to stabilize the material when high n-type densities are used (col. 2, lines 50 - 52 of Yoshida). One would be motivated to use silicon as

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an impurity element since silicon was a commonly used element in the semiconductor industry and is readily available with well-known properties, and one skilled in the art may adjust the band-gap and electrical properties of the semiconductor to using Si to optimize the semiconductor properties, such as conductivity, for a desired circuit, as discussed by Hasegawa (¶ [0037]).

b. Regarding claim 16, **Imai in view of Yoshida disclose a method of forming a diamond n-type semiconductor according to claim 15, as cited above, however they are silent with respect to explicitly disclosing that Si is artificially introduced as the impurity element to said diamond semiconductor substrate.**

Hasegawa discloses that silicon can be used as an impurity element when doping semiconductor diamond (paragraphs [0037] – [0038]).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the method of Imai in view of Yoshida, such that the other impurity element is silicon since Hasegawa discloses that silicon can be used in such concentrations to dope semiconductor diamond with a p-type material, and Yoshida discloses that a combination of n-type and p-type materials may be doped into a diamond semiconductor in order to stabilize the material when high n-type densities are used (col. 2, lines 50 - 52 of Yoshida). One would be motivated to use silicon as an impurity element since silicon was a commonly used element in the semiconductor industry and is readily available

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with well-known properties, and one skilled in the art may adjust the band-gap and electrical properties of the semiconductor to using Si to optimize the semiconductor properties, such as conductivity, for a desired circuit, as discussed by Hasegawa (¶ [0037]).

10. Claim 12 is rejected under 35 U.S.C. 103(a) as being unpatentable over Imai in view of Shiomi et al. (US 5,252,840, prior art of record). **Imai discloses a diamond n-type semiconductor according to claim 1, but is silent with respect to disclosing the device further comprises a second diamond semiconductor provided adjacent to said first diamond semiconductor and turned out to be n-type, wherein said second diamond semiconductor has an electron concentration exhibiting a negative correlation with temperature and a Hall coefficient not exhibiting a positive correlation with temperature, in the temperature range.**

Shiomi discloses that a second diamond semiconductor may be provided adjacent to a first diamond semiconductor (e.g. figure 1(b), second diamond semiconductor 3, disclosed in col. 9, lines 46 – 48, adjacent to first diamond semiconductor 2, disclosed in col. 8, lines 21 - 22),

wherein said second diamond semiconductor has an electron concentration exhibiting a negative correlation with temperature and a Hall coefficient not exhibiting a positive correlation with temperature, in the temperature range (col. 5, lines 16 – 24 disclose the structural characteristics of the device and layers. Since the device of Shiomi meets the structural limitations of the

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claimed invention of the Applicant, the properties of the applicant's invention, such as the temperature dependence of the electron concentration and Hall coefficient as claimed in claim 12, are presumed inherent to the device of Shiomi. See MPEP 2112.01).

It would have been obvious for one of ordinary skill in the art at the time the invention was made to modify the device of Imai such that a second diamond semiconductor layer was adjacent to the first diamond semiconductor layer, with the claimed properties, since Shiomi discloses that one may form such structures to achieve desired conduction properties of the device (col. 9, lines 61 – 68). One would have been motivated to form a second diamond semiconductor layer adjacent to the first diamond semiconductor layer in order to allow charge carrier diffusion from the first layer into the second layer, thereby altering the conduction properties of the device, as disclosed by Shiomi (col. 5, line 16 – 21).

Although Shiomi is silent with respect to the second semiconductor being n-type, Shiomi discloses the first and second diamond semiconductor layers to be p-type doped diamond semiconductor (col. 8, line 22 and col. 9, lines 48 - 49), and it is well-known in the art that one may interchange p-type and n-type doping to achieve a desired charge carrier concentration of either holes or electrons (e.g. as discussed in Shiomi, col. 1, lines 51 – 55, and Imai, col. 1, lines 16 - 24). One would have been motivated to substitute n-type doping for p-type doping in the first and second layers of Shiomi in order to create an n-type device, which would allow

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one to form complimentary circuits well-known in the semiconductor art (e.g. pn junctions).

Response to Arguments

11. Applicant's arguments filed June 2, 2009 have been fully considered but they are not persuasive.

a. With respect to the rejection of claims 1 – 11 and 13 – 16 under 35 U.S.C. 112, first paragraph, the Applicant argues that the specification is enabled, and that the rejection should be withdrawn. The Examiner respectfully disagrees. As cited above with respect to the rejection, independent claims 1 and 15 recite the property of the device such that there exists “*an electron concentration of said first diamond semiconductor exhibits a negative correlation with temperature, in a temperature range having a width of 100°C or more and included within a temperature region from 0°C to 300°C*”. However, it is well-known in the art that carrier concentrations (i.e. electron concentration in the valence bands of the semiconductor device) exhibit a positive, monotonic correlation with temperature. The Applicant has not provided adequate support and reasoning in the specification for the claim to the property of a carrier concentration that correlates negatively with temperature, which is an apparent violation of known physical law. Without proper support for such a property that is in direct violation of known physical law, the Examiner maintains that one of ordinary skill in the art would not

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know how to properly make and use the device. Hence, the specification is deemed not to be enabled properly.

b. With respect to the Applicants argument that the prior art of Imai fails to teach "*a first diamond semiconductor which has an n-type conduction and in which a distortion is artificially formed*", as cited in claims 1 and 15, the Applicant argues that the Examiners prior choice of the phrase "*a distortion may be formed in the lattice due to the sulfur impurity*" (Office Action filed March 2, 2009) implies that the is not necessarily a distortion formed, and hence that inherency is not shown with respect to the rejection of claims 1 and 15 over the prior art Imai. The Examiner has clarified the rejection above such that the comment reads "*a distortion is formed in the semiconductor lattice due to the sulfur impurity*". It is well-known in the art that an impurity doped into a lattice of a different elemental structure will necessarily form a distortion due to the difference in both atomic size (each element has a different atomic size) and different binding energy of the impurity to the surrounding lattice. Hence, the doping of sulfur into the diamond (carbon) lattice of Imai will necessarily produce a distortion of the carbon lattice surrounding the sulfur impurity.

c. With respect to claim 15 and the Applicant's argument that the prior art of Yoshida teaches away from the artificial formation of a distortion, it is respectfully submitted that Yoshida is not used to teach an artificial formation of a distortion

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within the diamond semiconductor. As cited above, the prior art of Imai discloses the artificial formation of the distortion of the diamond semiconductor. Yoshida is used to teach an introduction of an impurity element other than a donor element into the diamond semiconductor.

Furthermore, it is not found that Yoshida teaches away from artificially forming a distortion in the semiconductor. As cited in col. 2, line 52 of Yoshida (and page 10 of Applicant's Remarks), the diamond semiconductor may be simultaneously doped with p-type and n-type dopants, which are disclosed in co. 5, lines 59 - 61 to be Phosphorus and Hydrogen into the diamond crystal semiconductor. As discussed above in section (b) of the Response to Arguments, the introduction of a dopant into an lattice of different elemental structure necessarily forms a distortion of that lattice due to different atomic sizes of the dopant atom and lattice atoms, as well as due to the different binding energies between the dopant atom and surrounding lattice atoms. Hence, Yoshida teaches inherently a distortion of the diamond semiconductor. Since the doped semiconductor is formed by introducing the dopant atoms into a diamond film, one may consider the distortion to be artificially formed.

d. With respect to claims 9, 10, and 16, and in response to applicant's argument that there is no suggestion to combine the references, the examiner recognizes that obviousness can only be established by combining or modifying the teachings of the prior art to produce the claimed invention where there is

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some teaching, suggestion, or motivation to do so found either in the references themselves or in the knowledge generally available to one of ordinary skill in the art. See *In re Fine*, 837 F.2d 1071, 5 USPQ2d 1596 (Fed. Cir. 1988) and *In re Jones*, 958 F.2d 347, 21 USPQ2d 1941 (Fed. Cir. 1992). In this case, claims 10 and 16 recite the doping of silicon as a (second) impurity element into the semiconductor substrate. The Examiner has provided the teaching of Yoshida to disclose the doping of multiple elements into a diamond semiconductor substrate, and has used the teaching of Hasegawa to disclose that silicon may be used as an impurity dopant in a semiconductor substrate, with the motivation that one skilled in the art may adjust the band-gap and electrical properties of the semiconductor to using Si to optimize the semiconductor properties, such as conductivity, for a desired circuit, as discussed by Hasegawa (¶ [0037]). It is maintained that one of ordinary skill in the art would be able to combine the references of Yoshida, Imai, and Hasegawa to render obvious the claimed invention of claims 9, 10, and 16 since all of the references are directed to a diamond semiconductor and the doping thereof.

e. The Examiner notes that with respect to independent claims 1 and 15, that the Applicant as cited the limitation “*an electron concentration of said first diamond semiconductor exhibits a negative correlation with temperature, in a temperature range having a width of 100°C or more and included within a temperature region from 0°C to 300°*”, but has not claimed substantial structural

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limitations to support the claimed properties. As per MPEP 2112.01, it has been held that when the prior art discloses the structure of the claimed invention, a prima facie case of anticipation or obviousness of the properties of the device, such as the correlation of the electron concentration with temperature, has been established. As cited above with respect to claims 1 and 15, the prior art of Imai, and Imai in view of Yoshida, discloses the structural limitations of the claim, and therefore the properties of the device are presumed inherent to the structures of Imai, and Imai in view of Yoshida. There has not been any structural limitations to overcome the anticipation or obviousness of the prior art, and hence the rejections of claims 1 and 15 over the structure of Imai, and Imai in view of Yoshida, are maintained.

Conclusion

. Applicant's amendment to claims 1 and 15 necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any

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extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to ROBERT HUBER whose telephone number is (571)270-3899. The examiner can normally be reached on Monday - Thursday (9am - 6pm EST).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Thao Le can be reached on (571) 272-1708. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Lex Malsawma/
Primary Examiner, Art Unit 2892

/Robert Huber/

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August 29, 2009